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Chemistry 100 Practice Final Exam
1 Chem 100 Practice Final Exam 1. Which one of the following numbers contains 4 significant figures? a. 6.43 b. 39.4 c. 0.04840 d. 0.0257 e. none of the above 2. In which of the following pairs of numbers does each member of the pair contain the same number of significant figures? 3. A balance has an accuracy of ±0.01 grams.

Chem 100 Practice Final Exam(1).pdf - Chem 100 Practice ...

Chemistry 100 Practice Final Exam Instructions: Do not begin the exam until you have been instructed to do so. You have 120 minutes to complete this exam. There are 50 multiple choice questions. You must use a number 2 pencil. You may use a scientific calculator. Make sure that you have written your name legibly on the scantron form.

Chemistry 100 Practice Final Exam Instructions

Chemistry 100 Final Exam (ALL SECTIONS) Multiple Choice Questions (12 points, 2 each): answer the following multiple choice questions by circling the best response in each case. 1. The Cl-C-Cl bond angle in the COCl₂ molecule is approximately: 92o, 104.5o.

Chemistry 100 Final Exam (ALL SECTIONS)

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Chemistry 100 Final Exam (ALL SECTIONS)

Chem. 100 Exam II (Practice Test) ATOMIC STRUCTURE: The following information applies to subatomic particles. If the correct answer is Proton, mark "A," neutron, mark "B," electron, mark "C." If the answer is correct for more than one, mark all correct answers. 1. The mass of the particle doesn't weigh enough to worry about. 2.

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General Chemistry 1 2018: Sample Exams and Exam Solutions

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Chemistry 100 Practice Final Exam Instructions

Chemistry 100 Exam 1 Roadmap to Chemistry Mass Calculations Grams of element Molar mass of element (g/mol) Moles of element $\text{SYRJDGUR}^{\text{IV}}$ number of atoms or ions Atoms or Ions Atomic mass (amu / atom) Amu of element Grams of compound A Molar mass of compound A Moles of compound A $\text{SYRJDGUR}^{\text{IV}}$ number of molecules Molecules Formula mass (amu / molecule)

Instructor: Nicole Stevens 1 of 8 Winter 2008 Chemistry ...

Chemistry 11 I. Conserving significant figures. 0.0250 m is equivalent to: A. 0.25 cm B. 2.5 cm C. 2.50 cm D. 25.0 cm E. none of the above 2. In which of the following sets of quantities

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Organic Chemistry 2 Final Exam Review Multiple Choice Test ...

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