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Three States of

Matter \ " The

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States of matter

| States of

matter and

intermolecular

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Changes of

States of Matter

Chemical and

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Gas *10 Amazing*

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Pure Substances
and Mixtures!
(Classification
of Matter)

Physical and
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Understanding
Atoms, elements,
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of Matter

Physical and

Chemical

Changes:

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Changes States

of Matter –

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States of Matter

and Changes of

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\u0026 Changing

State #20 Pure

Substances and

Mixtures,

Elements \u0026

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Classification

of Matter,

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Ozone Layer,

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atmosphere,
ozone formation,
chlorofluorocarb
ons, CFC's 1.2

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Matter Chemistry
Central Science

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Chapter 1

Section 2 MATTER

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Chemistry

and Its And

Properties Matter

- Anything that

has MASS and

VOLUME. ATOMS

and MOLECULES

the basic

building blocks

of matter.

ATOM-- the

smallest unit of

an element that

still maintains

the properties

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that element.

ELEMENT- a pure substance made up of only one type of atom.

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about atomic
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radius. Circle
the letter of
the choice that
best completes

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the statement or
answers the
question. 1.
Atomic radii
cannot be
measured ...

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ter-Properties

and Changes

Solutions Manual

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3.35 Section 3.1

Properties of
Matter pages
70-75 Problem-

Solving Lab 1.

Explain why the
flow of a
compressed gas
must be
controlled for
practical and
safe use.

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Table and

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Section 6.1

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1. octaves 2.

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eight 3. nine 4.

accepted 5.

Dmitri Mendeleev

6. atomic mass

7. elements 8.

atomic number 9.

Henry Moseley

10. protons 11.

periodic law 12.

properties 13.

14 ...

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5 25 Electrons

in Atoms

Reviewing

Vocabulary Match

the definition

in Column A with

the term in

Column B. Column

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A Column B 1.

The set of frequencies of the

electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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numbers used in

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q... the study of

energy changes

(that is, energy

produced or

absor... A system

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extensive

property.

intensive

property.

extensive

properties

(examples)

intensive

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Answers

(examples) a
property that
depends on the
amount of matter
in a sample. a
property that
depends on the
type of matter
in a sample
not... mass,
volume, length,
weight.

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11 31 number of

moles of known

substance C H C

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H C H O O CO
CO CO H O H O H
O number of
moles of unknown
substance O CO H
O C H CO H O C H
O H O C H O CO
use mole ratio 5
mol O 2 1 mol C
3 H 8 3 mol CO 2
1 mol C 3 H 8 4
mol H 2 O 1 mol
C 3 H 8 1 mol C
3 H 8 5 mol O 2

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MASTER 16

A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or

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more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of
Compounds Quite
different
properties than
their component
elements.

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