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Finding  
Molarity, Mass  
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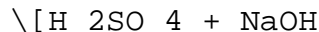
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Q2. Q3. Q4.

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Q1. Given the  
following  
reaction:



...

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Stoichiometry  
worksheet 1

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co 2 c. Answer  
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C 4h 10 co 2 e.

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following

equation. Given

the following

equation. C 4h

10 h 2o 2. O<sub>2</sub>

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h<sub>2</sub>o d. 2 will be  
formed from 1.65  
moles of kclo.

How many moles  
of o. C<sub>4</sub>H<sub>10</sub> CO<sub>2</sub>  
e. Stoichiometry  
worksheet 1

answers. 2 C<sub>4</sub>H<sub>10</sub>  
13 O<sub>2</sub> 8 CO<sub>2</sub> 10

H<sub>2</sub>O show what  
the following  
molar ratios  
should be.

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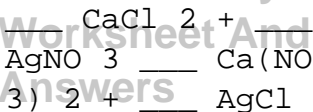
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Directions:

Solve each of  
the following  
problems. Show  
your work,  
including proper  
units, to earn

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full credit. 1.



How many grams  
of silver  
chloride are  
produced when 45  
g of calcium  
chloride react  
with excess  
silver nitrate?  
116 g AgCl 1 mol  
AgCl

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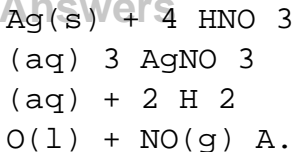
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Directions:

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the following  
problems. Show  
your work,  
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units, to earn  
full credit. 1.  
Silver and  
nitric acid  
react according

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to the following  
balanced  
equation: 3



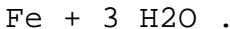
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of  $\text{LiNO}_3$ ) 4)



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Using the  
following  
equation:  $\text{Fe}_2\text{O}_3$   
+ 3  $\text{H}_2$  -----> 2



Calculate how  
many grams of  
iron can be made  
from 16.5 grams  
of  $\text{Fe}_2\text{O}_3$  by the  
following  
equation.

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Stoichiometry.  
Part 1: Mole ??  
Mass  
Answers.

Convert the  
following number  
of moles of  
chemical into  
its  
corresponding  
mass in grams.

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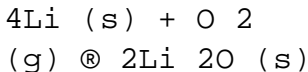
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Practice: 1. In  
the reaction:



a. what is the  
product? b. what  
are the  
reactants? c.  
what does the

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“(s)” after the  
formula of  
lithium oxide  
signify?

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reactants  
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an experiment 3  
25 g of nh<sub>3</sub> are

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allowed to react  
with 3.50 g of  $\text{O}_2$ .  
 $\text{NH}_3$  is the limiting  
reactant.

Reactant

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KClO<sub>3</sub> 3 mol KClO  
3 mol O<sub>2</sub> = mol O  
2 3.50mol KCl =  
mol KClO<sub>3</sub> 3 =  
0.275 mol Fe =  
mol Fe 20 3 = =

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the elements ...

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Using the  
following

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equation:  $\text{Fe}_2\text{O}_3$



Calculate how many grams of iron can be made from 16.5 grams of  $\text{Fe}_2\text{O}_3$  by the following equation.

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Part 1: Mole ??

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Convert the following number of moles of chemical into its corresponding mass in grams.

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