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Properties of the Elements
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Properties of the Elements:
Part 1 of 11

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Quantum Numbers, Atomic
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The Periodic Table: Atomic
Radius, Ionization Energy,
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Inside the Cell Membrane *How
to Write the Electron
Configuration for an Element
in Each Block*

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thing. Terms in this set
(25) Chemical formula.
indicates the relative
number of atoms of each kind
in a chemical compound.
Reveals the number of atoms
of each element contained in
a single molecule ...

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CHAPTER 7 REVIEW Chemical
Formulas and Chemical
Compounds SECTION 1 SHORT
ANSWER Answer the following

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Review Answer Key
questions in the space
provided. 1. c In a Stock
system name such as
iron(III) sulfate, the Roman
numeral tells us (a) how
many atoms of Fe are in one
formula unit. (b) how many
sulfate ions can be attached

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to the iron atom. (c) the charge on each Fe ion.

*7 Chemical Formulas and
Chemical Compounds*

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chemistry chapter 3. 30
terms. kagura_156 ...

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CHAPTER 7 REVIEW Chemical
Formulas and Chemical

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Compounds MIXED REVIEW SHORT
ANSWER Answer the following
questions in the space
provided. 1. Write formulas
for the following compounds:
CuCO₃ a. copper(II)
carbonate Na₂SO₃ b. sodium
sulfite (NH₄)₃PO₄ c.

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ammonium phosphate SnS_2 d.
tin(IV) sulfide HNO_2 e.
nitrous acid 2.

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Homework Answer Key

CHAPTER 5 REVIEW The
Periodic Law SECTION 1 SHORT
ANSWER Answer the following
questions in the space
provided. 1. c In the modern
periodic table, elements are
ordered (a) according to

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decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

2. d Mendeleev noticed that certain similarities in the

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5 The Periodic Law

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the space provided 9 1 1012
m What is the wavelength of

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light that has a frequency
of 3×10^4 Hz in a vacuum ...

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