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each reaction below, identify the atom
oxidized, the Page 5/21. Read PDF Oxidation
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reduced, the oxidizing agent, and the

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reducing agent. 1) $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$ 2)
 $2\text{Fe} + 3\text{V}_2\text{O}_3 \rightarrow \text{Fe}_2\text{O}_3 + 6\text{VO}$ 3) $2\text{KMnO}_4 + 5\text{KNO}_2 +$
 $3\text{H}_2\text{SO}_4 \rightarrow 2\text{MnSO}_4 + 3\text{H}_2\text{O} + 5\text{KNO}_3 + \text{K}_2\text{SO}_4$
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